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textbook, that is.) E1-26 The room volume is $(21 \text{ m} \times 13 \text{ m} \times 12 \text{ m}) = 3276 \text{ m}^3$. The mass contained in the room is $(3276 \text{ m}^3)(1.21 \text{ kg/m}^3) = 3964 \text{ kg}$: E1-27 One mole of sugar cubes would have a volume of $N_A \times 1.0 \text{ cm}^3$, where N_A is the Avogadro constant.

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