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methylpropene) 3 90-150 (CH₂), 90
-150 (C₅ 40 H 3 2) propene 3) 90-150
(CH₂), 90 -150 (C₅ 40 H 3
2-chloropropane), 2 5-40 (CH₃ 10-70
(CHCl) propanone), 2 20-50 (CH₃
190-220 (CO) methylamine 1 25-60 (CH₃
3) ethyl propanoate 3 5 2 5-40 (CH CH₂

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SECTION 1 - ¹H NMR Why compounds
absorb radiowaves (background
information beyond specifications) NMR
(nuclear magnetic resonance) is a very

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powerful tool for identifying compounds.
H, The nucleus of some atoms has
nuclear spin (e.g. ^1C , ^{19}F , ^{31}P),
although many atoms do not have any
nuclear spin (e.g. ^{12}C).

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OH NO 2 OH NO 2 OH NO 2

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2-chloro-5-nitrobenzaldehyde

Isobutylbenzene m-xylene o-nitrophenol

m-nitrophenol p-nitrophenol o-vanillin

phenol. 7.4 7.2 7.0 6.8 6.6 6.4 6.2 6.0

5.8 5.6 5.4 5.2 5.0 5.1

1H#NMR'Spectroscopy'Worksheet' Part'I'

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Chemsheets Model 3: ^{13}C NMR

Spectroscopy – Number of Signals

1. The molecule on the left has 5 types of C atom and the molecule on the right has 7 types of C atom *.
2. The molecule on the left will give 5 signals and the molecule on the right will give 7 signals.
3. See below.

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Chemsheets relative intensity = 0.3 :

0.15 : 0.3 : 0.6 = 2 : 1 : 2 : 6 TASK 2 -

Finding the relative intensity of signals

from a spectrum For each of the NMR

spectra below, calculate the relative

number of H atoms associated with each

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signal. 8 7 6 5 4 3 2 1 0 chemical shift δ

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Propene reacts with HBr to form H. H reacts with sodium hydroxide to form I, and I reacts with warm acidified potassium dichromate (VI) to form J. The infra-red spectra of H, I and J are given below, but it does indicate which is -

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which. Identify the three compounds H, I and J, using the infra-red spectra below, and decide which spectrum belongs

IR TASK 1 - Weebly

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Experiment 8.17 - NMR Using Liquid-Crystal

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^1H and ^{13}C NMR Spectroscopy
Answers Mention of acid eg H_2SO_4
OR alkali eg NaOH loses both M2 and
M3. 1 M4 Nucleophilic substitution Can

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only score M3 if reagent correct. Ignore alcohol or ethanol (conditions) or Temp.
1 (c) Bromine (penalise Br but mark on)
Acidified KMnO_4 (Penalise missing acid but mark on) Wrong reagent = no marks.

^1H & ^{13}C NMR Spectroscopy answers

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Answer: d. NMR SPECTROSCOPY MCQs.
10. Nuclei having either the number of protons or neutrons as odd have _____ spin. a) Integral spin b) Half integral spin c) Zero spin d) Positive spin. Answer: b.
NMR SPECTROSCOPY Objective Questions with Answers. 11. If the number of protons or neutrons is even the spin of the nucleus will be which of

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© www.CHEMSHEETS.co.uk 17-Jul-12
Chemsheets A2 029 9 IR TASK 2 1) The
IR spectra of six compounds are shown.
The compounds are: butanoic acid
butanone but-3-en-1-ol

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The problems are chosen to
demonstrate the most common patterns
in ^1H NMR spectroscopy, as well as, the

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situations where you need to consider the possibility of signal overlapping, incorrect absolute values of integrations, as the instrument measures only the relative area for each peak, examples where fairly large molecules give rise to spectra with few signals because of the symmetry elements.

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